

1021 Main Street Winchester, MA 01890-1943 617-721-4000

May 19, 1993 Project 92294

Chief, Geotechnical Engineering Division U.S. Army Corps of Engineers, New England Division 424 Trapelo Road Waltham, MA 02254

Dear Sir:

Re: Report on Well Cleaning Activities
Hopkinton Lake Dam
Hopkinton, New Hampshire
Contract DACW-33-91-D-008, Delivery Order 7

This letter report describes the work performed in November 1992 regarding cleaning of relief wells at Hopkinton Lake Dam. GEI Consultants, Inc. (GEI) engaged Alford Roger Cullimore Concepts Inc. (ARCC) of Daytona Beach, Florida, to clean and redevelop the wells using the "Blended Chemical Heat Treatment" process. During initial bailing of the wells and removal of sediment, it appeared that the sediment in the wells was contaminated with creosote or other compounds. Because of concerns regarding potential contamination of the Contoocook River and disposal of contaminated sediment, the Corps of Engineers decided to suspend work.

Summary of Work

Thursday, October 29, 1992 - ARCC arrived at the site with rig.

Week of Monday, November 2, 1992 - ARCC prepared rig and equipment. Chemicals were delivered and unloaded.

Thursday, November 5, 1992 - Site meeting with GEI, ARCC, and Corps of Engineers personnel to discuss details of work and schedule.

Concord, New Hampshire Raleigh, North Carolina Denver, Colorado Carlsbad, California

Monday, November 9, Through Thursday, November 12, 1992 - ARCC and GEI began bailing wells using a pump to remove sediment. Observed possible creosote contamination. Obtained samples of sediment using a bailer for delivery to Corps and for chemical testing by Corps environmental laboratory.

<u>Friday, November 13, 1992</u> - ARCC reloaded chemicals for return to supplier and demobilized rig.

Wednesday, March 3, 1993 - GEI received Corps letter directing termination of Delivery Order No. 7 of Contract No. DACW-33-91-D-008.

Sediment Sampling

The well locations are shown on the attached sketch (Fig. 1) provided by the Corps. The design drawings for the wells (Fig. 2) indicate that the relief wells are 74 feet deep. The GEI measured depth to the top of sediment and estimated sediment thickness in each well was as follows. The Corps measurements from August 1992 are shown for comparison. Figures 3 and 4 show the Corps data of sediment thickness over time.

Well	Depth from Top of Casing to Top of Sediment Measured by GEI November 11, 1992 (feet)	Estimated Sediment Thickness (feet)	Sediment Thickness Reported by Corps in August 1992 (feet)	
RW-1	71.0	3.0	2.7	
RW-2	62.4	11.6	10.8	
RW-3	67.5	6.5	6.5	
RW-4	63.7	10.3	10.4	
RW-5	64.9	9.1	8.4	
RW-6	61.9	12.1	11.5	
RW-7	66.1	7.9	6.3	
RW-8	67.5	6.5	9.0	

Each of the eight wells was bailed with a 4-inch-diameter bailer to recover some of the sediment at the base of the wells. Material recovered consisted of black sludge/organic silt-like material with minor quantities of silt and fine to medium sand. The material

exhibited an oily sheen and had a strong smell similar to that of creosote. Additional samples recovered at RW-4 to a depth of about 70 feet indicated a similar composition at depth with the material perhaps being slightly denser and containing an increased quantity of fine to medium sand. No stratification was noted, but agitation of the sediment by the action of the bailer may have disturbed any stratification.

The liquid above the sediment was a black slurry, also with an oily sheen. Samples of the slurry and sediment from RW-4 and RW-6 were obtained for chemical analysis at the Corps environmental laboratory in Hubbardston, Massachusetts.

Results of Chemical Analyses

The results of chemical analyses performed by the Corps are presented in Appendix A. The Corps report concludes that the sediment obtained from the wells was contaminated with coal/oil tar creosote.

Sincerely yours,

GEI CONSULTANTS, INC.

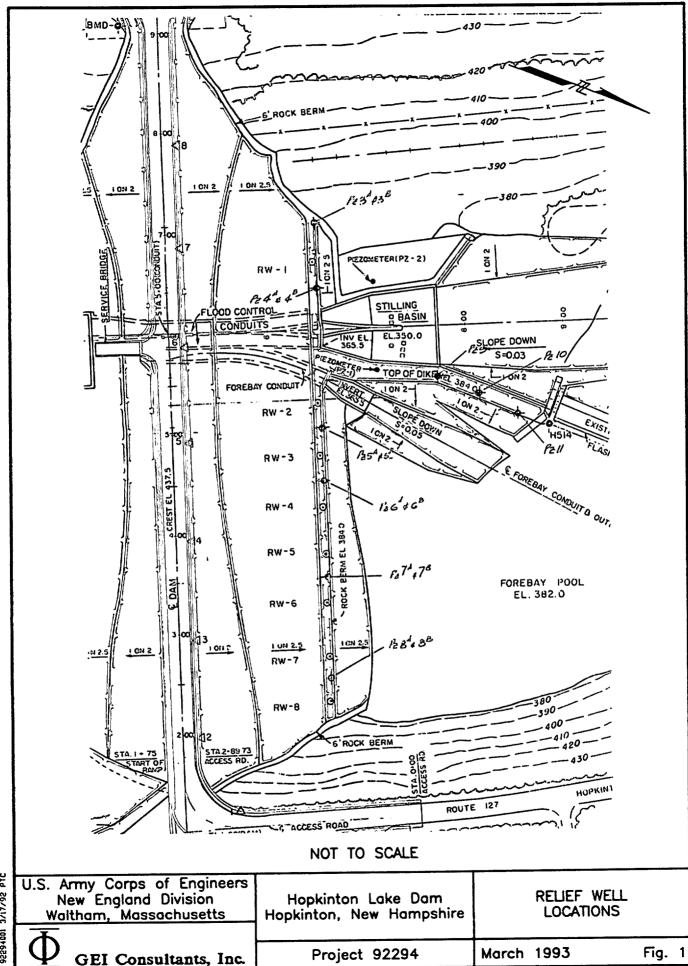
Michael Paster

Michael Paster, P.E.

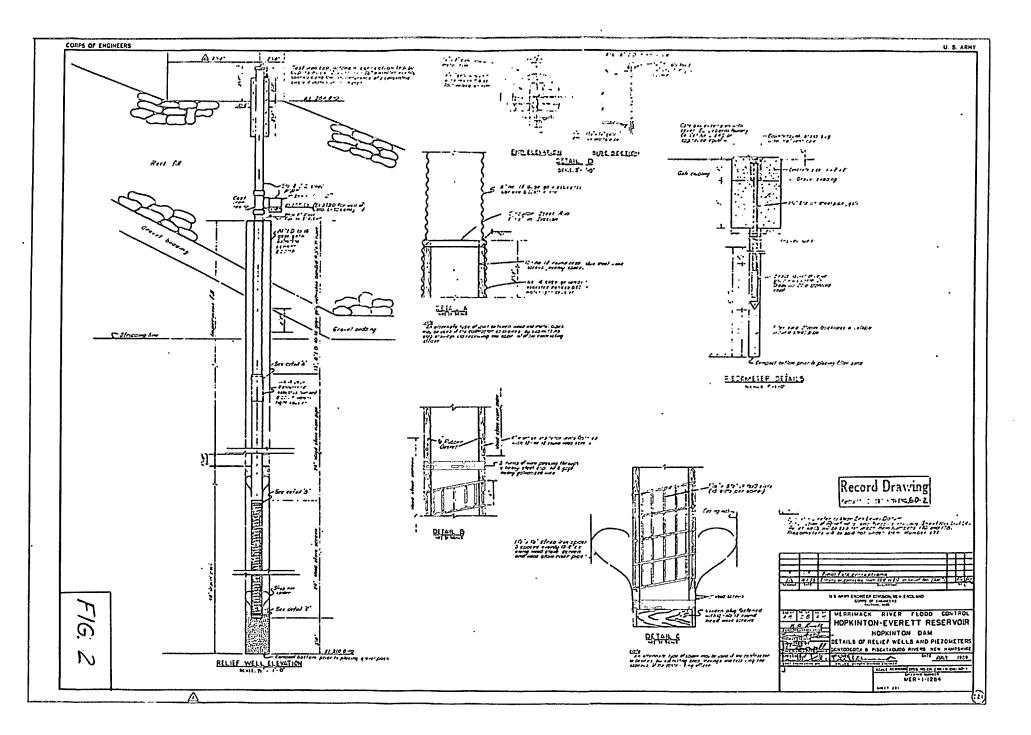
Project Manager

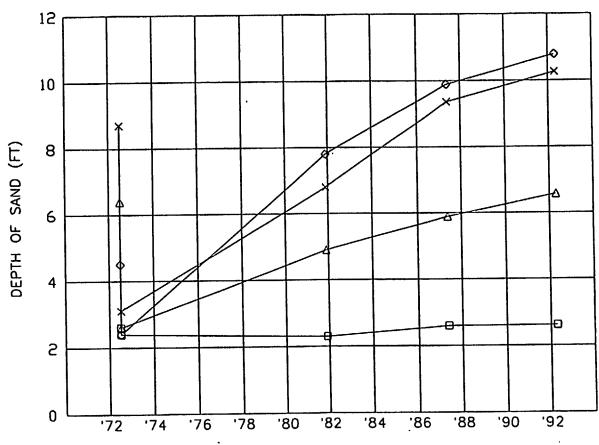
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Attachments



92294001 3/17/92 PTC





YEAR (19__)

—— RW-1 —→ RW-2 —— RW-3 —— RW-4

NOTE: The wells were flushed and cleaned in the summer of 1972

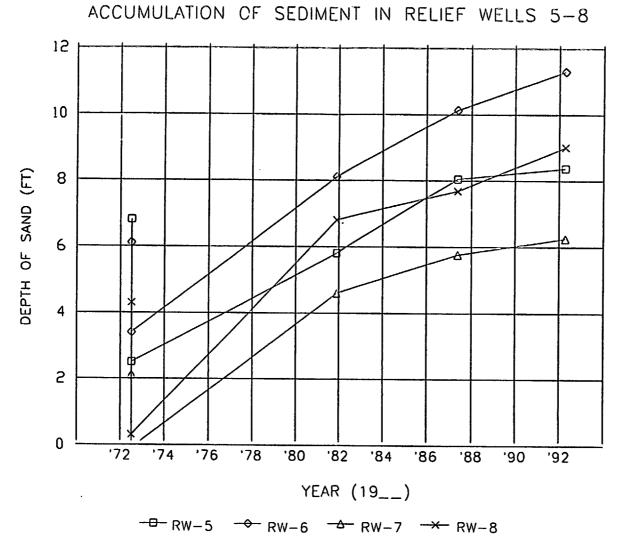
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MERRIMACK VALLEY FLOOD	CO

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GEOTECH. ENG. DIV. PLATE NO. 67 SCALE: AS SHOWN DATE: AUGUST 1992

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NOTE: The wells were flushed and cleaned in the summer of 1972

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MAV	ACCUMI	DPKINTON DAM JLATION OF SEDIMENT
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DATE: AUGUST 1992

APPENDIX A

Analytical Data Report Hopkinton Wells

by

U.S. Army Corps of Engineers New England Division Environmental Laboratory Hubbardston, MA 01452

March 26, 1993



Hubbardston MA 01452

Analytical Data Report

HOPKINTON WELLS

U.S. Army Corps of Engineers New England Division Environmental Laboratory Hubbardston, MA 01452

Date: March 26, 1993

Brian J. Condike

Chief, Environmental Laboratory

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1. Case Summary

HOPKINTON WELLS (11/13/92)

- 1. Two sediment samples were received for the above subject project on 13 November 1992. The correct sample containers and sample preservation procedures were followed unless otherwise indicated on the cooler receipt form. Copies of the chain-of-custody records are enclosed for reference, along with a list of the samples collected.
- 2. The following analyses were performed in-house:

<u>Analysis</u>	EPA Method

Sediment samples:

Fuel Identification	*
Total Petroleum Hydrocarbons	9071/418.1
Arsenic	3051/7060
Lead	3051/6010
Volatile Organics	8240
Polynuclear Aromatic Hydrocarbons	3540/8270

* - Proposed Practive Oil Spill Source Identification by Combined Gas Chromatography and Positive Ion Electron Impact Low Resolution Mass Spectrometry, ASTM, Draft 1, Jan., 1991.

Contaminated Soils - Diesel Fuel Contamination, written by Paul T. Kostecki and Edward Calabrese, Chapter 1 - The Use of Hydrocarbon Analyses for Environmental Assessment and Remediation.

2. Sample Listing

3. Laboratory Data

Samples 19445 and 19446 demonstrate a predominance of polynuclear aromatic hydrocarbons (PAH) and in particular the pyrogenic polynuclears fluoranthene, phenanthrene, and pyrene. The relative concentrations of the polynuclear aromatics for these environmental numbers are plotted in the accompanying figure along with the distributions for a coal-tar creosote, (all relative to chrysene). Aside from naphthalene, (which concentration could have easily been significantly reduced over years of water contact due to solubilization), the predominance of the pyrogenic PAHs for both samples and the creosote standard is evident.

Although coal tar/oil creosote does, in fact, have significant amounts of phenolic compounds, (about 10% by weight relative to 85% PAH content), since these acidic organics are extremely soluble in water, (5 to 6 orders of magnitude greater than the solubility of the PAHs), leaching over time could very easily, in the space of decades, reduce the phenolic concentrations drastically. (In fact only xylenol, one of the least soluble phenolics present in creosote was actually encountered near the detection limit).

Conclusion:

Samples 19445 and 19446 consist of coal oil/tar creosote. Creosote is commonly used to preserve wood to protect it from rot and worms. (Due to the relatively higher concentration of pyrogenic PAH's and lower concentrations of naphthalenes, these creosote samples are probably from a coal tar source). The fact that the phenolics are virtually absent can be accounted for by their solubilization/leaching over decades of years while the creosote was in water contact. Pentachlorophenol was not detected. Due to its relatively poor water solubility, it seems unlikely this chemical was added to the creosote for its wood preserving characteristics.

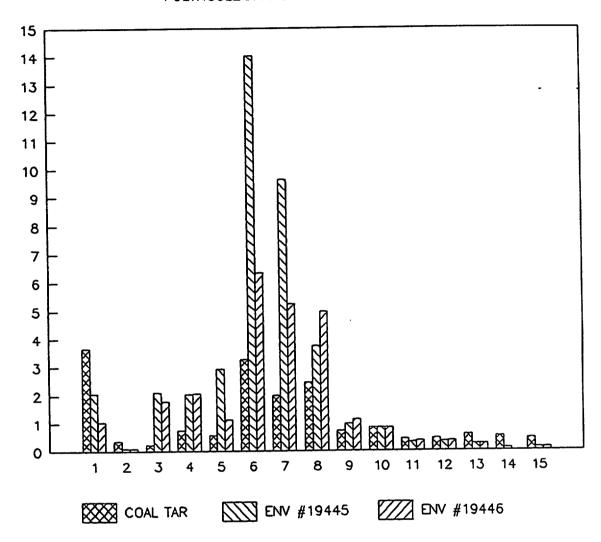
*Method References:

Proposed Practice Oil Spill Source Identification by Combined Gas Chromatography and Positive Ion Electron Impact Low Resolution Mass Spectrometry, ASTM, Draft 1, Jan, 1991.

Contaminated Soils - Diesel Fuel Contamination, written by Paul T. Kostecki and Edward Calabrese, Chapter 1 - The Use of Hydrocarbon Analyses for Environmental Assessment and Remediation.

RELATIVE CONCENTRATIONS OF PREDOMINANT

POLYNUCLEAR AROMATIC HYDROCARBONS



- 1 Naphthalene
- 2 Acenaphthylene
- 3 Acenaphthene
- 4 Fluorene
- 5 Anthracene
- 6 Phenanthrene
- 7 Fluoranthene
- 8 Pyrene
- 9 Benzo(a)anthracene
- 10 Chrysene
- 11 Benzo(b)fluoranthene
- 12 Benzo(k)fluoranthene
- 13 Benzo(a)pyrene
- 14 Indeno(1,2,3-c,d)pyrene
- 15 Benzo(g,h,i)perylene

		U.S. AI	RMY CORPS OF ENGINEERS - ENVIRONM	ENTAL LABORATORY	January 23, 1993
	Lab#	Field Description	Test	Result Units	Date Analyzed
	A-19445	RW-4	Total Petroleum Hydrocarbons	1400 ppm	12/14/92
	A-19446	RW-6	Total Petroleum Hydrocarbons	180 ppm	12/14/92
1 m /h					
		1/.0	A:16.		
	Reviewed B Approved B	y: Katherine P.	Miller, Analyst, Chief Chemist		
aris.					

PRODUCED ON

01/23/93

14:19

HOPKINGTON WELLS (11/13/92)

METHOD 9071/418.1: TOTAL PETROLEUM HYDROCARBONS (ppm) - SEDIMENT

ENV NO. TOTAL PETROLEUM DATE DATE HYDROCARBONS EXTRACTED ANALYZED

METHOD BLANK < 4.8 12/4/92 12/14/92

- TRACE METALS RESULTS

January 23, 1993

Lab#	Field Description	Test	Result	Units	Date Digested	Date Analyzed	
A-19445	RW-4	Arsenic Lead	28 130	ug/g ug/g	12/16/92 12/16/92	12/28/92 12/28/92	

Reviewed By: Approved By: Chief Chemist

- TRACE METALS RESULTS

January 23, 1993

Lab#	Field Description	Test	Result	Units	Date Digested	Date Analyzed
A-19446	RW-6	Arsenic Lead	11 70	ug/g ug/g	12/16/92 12/16/92	12/28/92 12/28/92

Reviewed By: Approved By:

Analyst

. Chief Chemist

- TRACE METALS RESULTS

January 23, 1993

Lab#	Field Description	Test	Result		Date Digested	Date Analyzed
A-19446	RW-6	Arsenic Lead	11 70	ug/g ug/g	12/16/92 12/16/92	12/28/92 12/28/92

PRODUCED ON

01/23/93 14:30

HOPKINTON WELLS (11/13/92)

TRACE METAL RESULTS - SEDIMENT (ppm)

,	PARAMETER	*	MET BLA		-	*
rsenic ead		*	< <	0.20 0.60		*

SAMPLE DATE:

DATE DIGESTED: 12/16/92
DATE ANALYZED: 12/28/92 - 1/5/93

02/02/93 07:53

PRODUCED ON

HOPKINGTON WELLS (11/12/92)

METHOD 8240: VOLATILE ORGANICS - SEDIMENT (ug/kg)

PARAMETER	*	***	19445 * RW-4 SEDIMENT		19446 * RW-6 SEDIMENT	***	METHOD BLANK SEDIMENT	
Chloromethane	•	<	25	<	25	<	5.0)
Vinyl chloride	•	<	25	₹	25	<	5.0	
Bromomethane	*	<	25	<	25	<	5.0	
Chloroethane	•	<	25	<	25	<	5.0	
1,1-Dichloroethene	•	<	10	<	10	<	2.0	
Acetone	•	<	150	<	150	<	30	
Carbon disulfide	*	<	10	<	10	<	2.0	j
Methylene chloride	•	В	46	8	128		5.1	ĺ
trans-1,2-Dichloroethene	•	<	10	<	10	<	2.0	j
1,1-Dichloroethane	•	<	10	<	10	<	2.0	1
cis-1,2-Dichloroethene	*	<	10	<	10	<	2.0	1
2-Butanone	•	<	150	<	150	<	30	
Chloroform	*	<	10	<	10	<	2.0)
1,1,1-Trichloroethane	*	<	10	<	10	<	2.0	j
_Carbon tetrachloride	•	<	10	<	10	<	2.0)
Benzene	*	<	10	<	10	<	2.0)
1,2-Dichloroethane	•	<	10	<	10	<	2.0)
Trichloroethene	•	<	10	<	10	<	2.0)
1,2-Dichloropropane	*	<	10	<	10	<	2.0)
Bromodichloromethane	*	<	10	<	10	<	2.0)
34-Hethyl-2-pentanone	*	<	100	<	100	<	20)
cis-1,3-Dichloropropene	•	<	10	<	10	<	2.0)
m Totuene	*	<	10	<	10	<	2.0)

PRODUCED ON

02/02/93

07:53

HOPKINGTON WELLS (11/12/92)

METHOD 8240: VOLATILE ORGANICS - SEDIMENT (ug/kg)

******	*****	***						************
PARAMETER	*	****	19445 * RW-4 SEDIMENT	***	19446 * RW-6 SEDIMENT	***	METHOD BLANK SEDIMENT	*
trans-1,3-Dichloroproper	ne *	<	10	<	10	<	2.0	•
1,1,2-Trichloroethane	•	<	10	<	10	<	2.0	*
Tetrachloroethene	*	<	10	<	10	<	2.0	•
2-Hexanone	*	<	100	<	100	<	20	•
Dibromochloromethane	•	<	10	<	10	<	2.0	•
Chlorobenzene	•	<	10	<	10	<	2.0	*
Ethylbenzene	•	<	10	<	10	<	2.0	•
m/p Xylene	•	j	10	<	10	<	2.0	•
0-Xylene	•	<	10	<	10	<	2.0	*
Styrene	*	<	10	<	10	<	2.0	•
8rcinoform	•	۲ .	10	<	10	<	2.0	•
1,1,2,2-Tetrachloroetha	ne *	<	10	<	10	<	2.0	*
**********	******	***	*******	***	******	****	******	*************
Surrogate Recovery	(%)							•
	(70-121)		116		114		90	*
Toluene D8	(84-138)		100		98		101	•
4-8romofluorobenzene	(59-113)	***	86		92		78	*
DILUTION FACTOR			5.0		5.0		1.0	*

SAMPLE DATE:

11/11/92

11/12/92

DATE ANALYZED:

11/19/92

11/19/92 11/19/92

⁻ Analyte also detected in the method blank.

⁻ Estimated value; analyte detected at < the Practical Quantitation Limit.

⁻ Highly degraded petroleum oil present.

PRODUCED ON

03/16/93 14:50

HOPKINTON WELLS (11/12/92)

EPA METHOD: POLYNUCLEAR AROMATIC HYDROCARBONS (ug/g)

ANALYTE	*	19445 RW-4	19446 RW-6	METHOD Blank	
	•	SEDIMENT	SEDIMENT	SEDIMENT	
Napthalene	*	118	22	< 0.0043	*******************************
2-Methyl napthalene	*	71	14	< 0.0043	
Acenaphthylene	*	5.0	1.8	< 0.0043	
Acenaphthene	*	121	38	< 0.0043	
Fluorene	*	117	44	< 0.0043	
Phenanthrene	•	U 795	135	< 0.0043	
Anthracene	*	167	25	< 0.0043	
Fluoranthene	*	545	111	< 0.0043	
Pyrene	*	212	105	< 0.0043	
Benzo(a)anthracene	*	56	24	< 0.0043	
Chrysene	*	48	18	< 0.0043	
Benzo(b)fluoranthene	*	19	8.4	< 0.0043	
Benzo(k)fluoranthene	•	20	8.9	< 0.0043	
Benzo(a)pyrene	•	16	5.7	< 0.0043	
Dibenzo(a,h)anthracene	*	J 1.5	< 0.081	< 0.0043	
Benzo(g,h,i)perylene	•	6.4	2.5	< 0.0043	
Indeno(1,2,3-cd)pyrene	*	5.2	< 0.081	< 0.0043	
*******	******	***********	********	*******	**********
•	•				
DILUTION FACTOR	•	1.1	0.81	0.043	
	*****	******	*******		
Surrogate Recoveries (%)	•				
	*				
2-Fluorobiphenyl (30-115)	*	92	91	89	
Nitrobenzene-D5 (23-120)	•	98	95	91	
* Terphenyl-D14 (18-137)	*	163	148	74	
************	*****	******	******	*****	************
J .					
SAMPLE DATE	:	11/11/92	11/12/92		
DATE RECEIVED	-	11/12/92	11/12/92		
77		7 - 7 - - 7 - - - -			

DATE RECEIVED: 11/12/92 11/12/92

DATE EXTRACTED: 11/24/92 11/24/92

DATE ANALYZED: 11/26/92 11/26/92 11/25/92

J - Estimate value; greater than Detection Limit, but less than Practical Quantitation Limit.

U - Above the upper calibration limit.

4. Quality Assurance Data

TOTAL PETROLEUM HYDROCARBONS BLANK SPIKE/BLANK SPIKE DUPLICATE

PRECISION	ľ
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100	COMPOUND DATE ANALYZED: 12/14/92	*	BLANK SPIKE RECOVERY (%)	*	BLANK SPIKE DUPLICATE RECOVERY (%)	* * *	RELATIVE PERCENT DEVIATION (RPD)	•	RPD Maximum	*	IN OR OUT OF QC LIMITS	•
*Tot	al Petroleum Hydrocarbons	*	55	*	65	****	17	*	66	****	IN	****

ACCURACY

	**********	*	BLANK	****	**** R	LANK	*	****	*****	*****	****	****	*	*****	***
_	* COMPOUND	*	SPIKE	•	_	ESULT	*	SPIKE	*	SPIKE	*	CONTROL	•	IN OR	*
	*	*	RESULT	*			*	ADDED	•	RECOVERY	*	LIMITS	*	CUT	ŧ
	*	*		*			•		*	x	*	REC	*	OF QC	*
۳	±	*		*			•		*		*		*	LIHITS	*
	******	***	*******	****	***	*****	******	******	*****	******	***	*****	****	*****	***
	*Total Petroleum Hydrocarbons	*	22	*	<	4.8	*	40	*	55	*	50 - 150	*	IN	*

TRACE METAL ANALYSIS BLANK SPIKE/BLANK SPIKE DUPLICATE SEDIMENT **PRECISION**

* Arsenic	PARAMETER 12/16/92	*	BLAHK SPIKE RECOVERY (%)	*	BLANK SPIKE DUPLICATE RECOVERY (%)	*	RELATIVE PERCENT DEVIATION (RPD)	*	MAXIMUM ACCEPTABLE RPD	* * *	IN OR OUT OF QC LIMITS	*
# Lead	Arsenic	*	94 101	***	91		3 4		30 30	*	IN .	*

TRACE METAL ANALYSIS BLANK SPIKE/BLANK SPIKE DUPLICATE SEDIMENT ACCURACY

	PARAMETER 12/16/92	*	BLANK SPIKE RESULT	* * *	BLANK RESULT	*	SPIKE ADDED	*	SPIKE RECOVERY	* *	CONTROL LIMITS REC	*	IN OR OUT OF QC LIMITS	*
	Arsenic	* 	4.7	_•	< 0.20	*	5.0	• :-	94	_•_	75 - 125	* 	IN	*
*	Lead	*	5.1	*	< 0.60	•	5.0	*	101	_*_	75 - 125	<u> </u>	IN	_*

VOLATILE ORGANIC BLANK SPIKE-BLANK SPIKE DUPLICATE SEDIMENT PRECISION

∰) •• •	********	*******	*****	*****	******	*****
	COMPOUND	* SPIKE * RECOVERY	BLANK SPIKE DUPLICATE RECOVERY	* RELATIVE * PERCENT * DEVIATION * (RPD)	* RPD	IN OR * OUT OF * QC * LIMITS *
	,1-Dichloroethene	• 71	* * 66 *	* * 7	* 22 *	• IH •
	Benzene :	80	* 81 *	* 1	* 21 *	* IN *
1	Trichloroethene	• 80 •	* 79 *	• 1 • 1	* 24 *	* IN *
* 1	Toluene	* * 71 *	* * 75 *	* * 5 *	* 21 *	* IN *
•	Chlorobenzene	* 74 *	* 79 *	* 7 *	* 21 *	+ IN +

SOIL ACCURACY

SPIKING COMPOUND	* MATRIX * SPIKE * RECOVERY	* ACCEPTABLE * RANGE *	* IN OR * * OUT OF * * QC * * LIMITS *
1,1-Dichloroethene	* 71 *	* 59 - 172 *	* * * * IN * * *
Benzene	* 80 *	* 66 - 142 *	• 1N •
Trichloroethene	* 80 *	*	* IN *
Toluene	* 71 *	* 59 - 139 *	* 1N *
Chlorobenzene	* 74 *	* 60 - 133	* IN *

POLYNUCLEAR ARMOMATIC HYDROCARBONS BLANK SPIKE-BLANK SPIKE DUPLICATE

SEDIMENT PRECISION

********	****	*****	*******	********	********
1	* BLANK	* BLANK	* RELATIVE	* MAXINUM	* IN OR *
СОМРОИНО	* SPIKE	* SPIKE	* PERCENT	* ACCEPTABLE	* GUT OF *
<i>}</i>	* RECOVERY	* DUPLICATE	* DEVIATION	* RPD	* QC *
DATE EXTRACTED: 11/24/92	*	 RECOVERY 	* (RPD)	•	* LIMITS *
************	*******	**********	**********	******	*****
	•	•	•	•	*
Acenaphthene	* 97	* 99	* 2	* 19	* 1N *
	• 63	* 64		* 36	* IN *
Pyrene	• 65	* 04	*	+ 30	. 14
					

ACCURACY

**********	*****	****	*****	*****	****	****	***
SPIKING	*	BLANK	•	ACCEPTABLE	•	IN OR	•
COMPOUND	*	SPIKE	•	RANGE	•	OUT OF	*
	*	RECOVERY	*		*	QC	*
	•		*		*	LIMITS	*.
************	*	******	****	*****	****	*****	****
Acenaphthene	*	97	*	31 - 137	*	IN	•
			_;-		-:-		—;
Pyrene	•	63	*	35 - 142	*	IN	*
	*		*		*		*

5. Chain of Custody

OE! Consultants, Inc. 53 Regional Drive Concord, NII 03301-8500 CHAIN OF CUSTODY RECORD DOCUMENT NUMBER PROJECT NAME AND NUMBER! HOPKINTON DAM 92294 SAMPLERS! NO. OF R. MAUCK REMARKS CON-TAINERS DATE (TIME) COMP CHAB STATION NO. SAMPLE NUMBER RW-6 1/1/92 130 5 RELINQUISHED, BY (SIGNATURE) DATE/TIME: (hours) RECEIVED BY (SIGNATURE) **RELINQUISHED BY (SIGNATURE)** DATE/TIME: RECEIVED BY (SIGNATURE) 11/13/92 Dar Josine Mecan NELTHOUISTED BY (SIGNATURE) DATE /TIMI: DECEIVED BY (SIGNATURE) RELINOUISHED BY (SIGNATURE) DATE/TIME RECEIVED BY (SIGNATURE): Joann Mercin 11/13/92 0853 HELINOUISHED BY (SIGNATURE) DATE/TIML DATE/TIML: RECEIVED FOR LABORATORY BY DATE/TIME: REMARKS:

1445

1446

6. Cooler Receipt Form

CENED-ED-GL-E SAMPLE CONTAINER RECEIPT FORM

ROJ	ECT: Hopkington Dam			
ont	ainer received on 11/13/92 and opened on 11/16/92 by:			
				
•	Shipper (USM, UPS, DHL, FEDEX, P/C, AIR EXP, HAND-DELIVERED)			
•	Container type (Cooler, box, envelope, etc.)		•	
•	Were custody seals on outside of container? How many & where:, seal date:, seal name:	1/A ——	Yes	Мо
•	Were custody papers taped to lid inside container?	A/N	Yes	No
•	Custody papers properly filled out? (ink, signed, etc.)		Yes	Мо
	Was project identifiable from custody papers?		Yes	Ио
•	Did you sign custody papers in appropriate place?		Yes	No
•	Did you attach shipper's packing form to this form?	A\N	Yes	Ио
•	Packing material (peanuts, vermiculite, bubble wrap, paper, ca	ans,	other	5)
Ο.	Was sufficient ice used? Temperature 3.8 °C	N/A	æs)	No
1.	Were all samples sealed in separate plastic bags?	N/A	Yes	No
2.	Did all samples arrive in good condition?		(Yes)	Мо
3.	Sample labels complete? (#, date, analysis, preservation, sig	n.)	(Yes)	Ио
4.	Did all sample labels agree with custody papers?	(Yes	Ио
5.	Were correct sample containers used for tests indicated?	N/A	Yes	Мо
6.	Were correct preservatives used? (TM pH, CN- pH)	N/A)	Yes	Мо
7.	Were VOA vials bubble-free (H2O) or no headspace (soil)?	N/A	Yes	NO
8.	Was sufficient amount of sample sent in each container?		Yes	Ио
9.	Were air volumes noted for air samples?	N/A	Yes	Ио
ο.	Were initial weights noted for pre-weighed filters?	Ń/A	Yes	Ио
is	crepancies:			

7. Quality Assurance Review

Quality Assurance Review Project: Hopkinton Wells

Date: March 22,1993

Soil #Samples-2 #Parameters-6

A. Sample Handling:

The samples were collected by a contractor using the appropriate sample containers and preservation techniques. The samples were not placed in separate bags. However, no problems were noted regarding this discrepancy. The proper chain of custody procedures were followed except that no custody seals were used. The samples arrived in good condition and were determined not to have been disturbed between collection and delivery to NED.

B. Laboratory Analysis:

1. Holding Times:

The maximum holding times between sample collection and analysis were met for arsenic, lead, and VOAs. The 28 day holding time for TPH was exceeded by five days. This will not be significant. The 14 day maximum holding time between sample collection and extraction was met for semi-volatiles as well as the 40 day holding time between extraction and analysis.

2. Method Blanks:

The method blanks for TPH, arsenic, lead, and PAH were free from contamination. A trace of methylene chloride was found in the VOA blank. It is not a problem in the low concentrations we experience. The presence of this common contaminant has been greatly reduced over the past year.

3. Methodology:

Standard EPA test procedures were used for all parameters except for fuel identification where references to a proposed ASTM procedure and a procedure from "Contaminated Soils- Diesel Fuel Identification" were used. They represent the state-of- the art methodology.

4. Surrogate Recoveries:

The surrogate recoveries for the semi-volatiles were in control except for two instances where the sample recoveries for terphenyl-d14 exceeded the acceptable range because of matrix effects. However, the departures not excessive. There may be some positive bias , but it will not be enough to affect interpretation of the sample data.

5. QA/QC Data:

The blank spikes and blank spike duplicates for the metals, TPH, VOAs, and PAHs were all in control for both precision and accuracy.

6. Assessment of data: The data appear reasonable and internally consistent.

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